



CATALYTIC DESTRUCTION OF AROMATIC VOCs ON SCR-DeNOx COMMERCIAL CATALYST

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Abstract

Aromatic volatile organic compounds (VOCs) proved to have enhanced polluting behavior upon the atmosphere and human health. Their catalytic destruction is an alternative for environmental pollution reduction in the case of gaseous emissions with low contents of VOC.

In this paper, an experimental study is presented, concerning the conversion of benzene and its mono-halogenated derivatives on SCR-DeNOx (V_2O_5 - WO_3 / TiO_2) commercial catalyst, specific for the reduction process of nitrogen oxides. The catalytic oxidation was carried out at temperatures ranged between $423\text{ }^{\circ}\text{K}$ – $623\text{ }^{\circ}\text{K}$ in different reaction environments (air, water vapors and halogenated acids) and for low concentrations of aromatic organic pollutants (< 100 ppm). The catalytic conversion of aromatic organic volatile compounds increases with temperatures and is influenced by the presence of water vapors and halogenated acids. It can reach values of above 75%, for a temperature of $623\text{ }^{\circ}\text{K}$. The reaction of catalytic oxidation is incomplete and leads to CO formation.

Keywords: catalytic destruction, benzene, halogenated derivatives, SCR-DeNOx catalyst

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